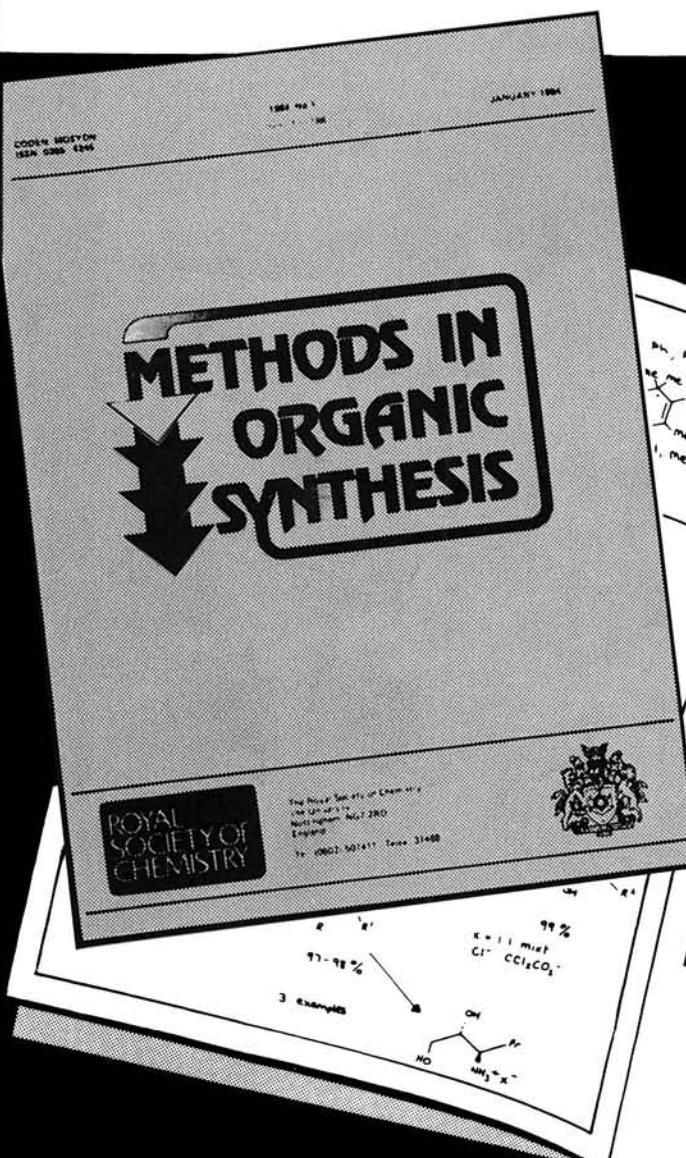


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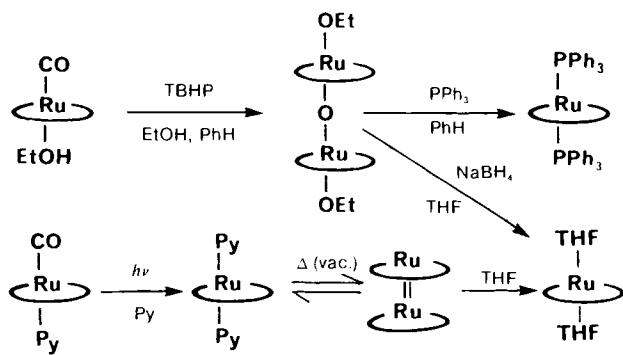
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CARBONYL UPDATE

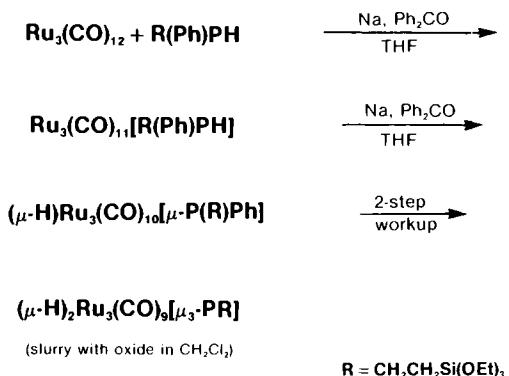
Clusters, Catalysts and Porphyrins

Carbonyl complexes of ruthenium(II) octaethylporphine (OEP) and tetraphenylporphine (TPP) are the latest additions to a growing family of Aldrich carbonyl compounds. The novel chemistry of Ru(II)(CO)OEP and Ru(II)(CO)TPP has only recently come under investigation. Their conversion to bis-ligand complexes or binuclear ruthenium(II) species is relatively straightforward.^{1,2}



The reactivity of the complexes toward small gas molecules (especially oxygen) is well documented,^{3,4} as is their catalytic activity in oxidation and decarbonylation reactions.^{5,6} Further derivatization to tertiary phosphine species,⁷ ruthenium(III) systems^{8,9} and oxo-ruthenium(IV) cation radical intermediates¹⁰ has been carried out. NMR studies of such π-cation radicals as peroxidase models have been reported.

The catalytic behavior of the ruthenium carbonyl cluster,¹² Ru₃(CO)₁₂, has been exploited in the reductive carbonylation of aromatic nitro compounds to carbamates.¹¹ In addition, a phosphine-stabilized ruthenium carbonyl cluster has been tethered to oxide supports, following the scheme below.¹⁴



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References:

- 1 Collman, J.P. *et al.* *J. Am. Chem. Soc.* **1981**, *103*, 7030.
- 2 Collman, J.P. *et al.* *ibid.* **1984**, *106*, 5151.
- 3 Farrell, N.P.; Dolphin, D.; James, B.R. *ibid.* **1978**, *100*, 324.
- 4 James, B.R. *et al.* In "Fundamental Research in Homogeneous Catalysis"; Tsutsui, M., Ed.; Plenum Press: New York, **1979**; Vol. 3, p. 751.
- 5 Domazetis, G. *et al.* *Am. Chem. Symp. Ser. No. 152*, **1981**, 243.
- 6 James, B.R. *et al.* *Inorg. Chim. Acta* **1984**, *85*, 209.
- 7 Ariel, S. *et al.* *Can. J. Chem.* **1984**, *62*, 755.
- 8 James, B.R. *et al.* *ibid.* **1984**, *62*, 1238.
- 9 Barley, M. *et al.* *ibid.* **1983**, *61*, 2389.
- 10 Leung, T.; James, B.R.; Dolphin, D. *Inorg. Chim. Acta* **1983**, *79(B7)*, 180.
- 11 Morishima, I.; Takamuki, Y.; Shiro, Y. *J. Am. Chem. Soc.* **1984**, *106*, 7666.
- 12 Lavigne, G.; Kacsz, H.D. *ibid.* **1984**, *106*, 4647.
- 13 Cenini, S. *et al.* *Chem. Commun.* **1984**, 1286.
- 14 Cook, S.L.; Evans, J. *ibid.* **1983**, 713.
- 15 Naito, S.; Tamaru, K. *Chem. Lett.* **1982**, 1833.

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29,202-8	Ru(II)(CO)OEP	100mg \$45.00; 500mg \$175.00
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